

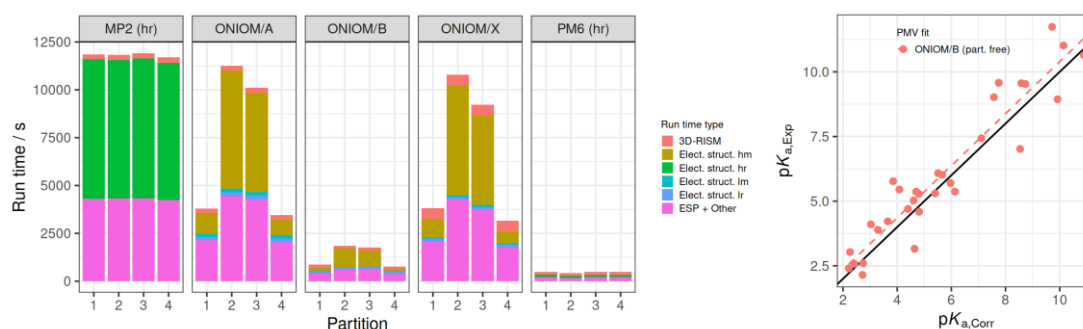
Accurate prediction of acidity constants with an ONIOM embedded cluster RISM approach

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The accurate prediction of physicochemical properties such as acidity constants (pK_a) plays a decisive role for, e.g., modelling protein function or in the development of drug-like molecules. As cellular environments show a variety of different pH values, the pK_a of a titratable residue or drug therefore heavily influences the mode and strength with which a compound may bind to its target. Our previous strategy for predicting pK_a values for small molecules focused on modelling the compounds' thermodynamics in an aqueous environment by means of our "embedded cluster RISM" (EC-RISM) solvation model and first-principles quantum mechanical (QM) calculations. [1] However, the size of most biomolecular systems prevents the application of these QM methods, thus prohibiting the accurate and granular modelling of the solvent environment with EC-RISM. Since the first suggestion by Warshel and Levitt [2] multiscale methods have emerged as an effective tool to model large-scale chemical processes in various environments, thus offering a route to expand the range of system sizes that can be modelled via EC-RISM theory.



Here we present a novel multiscale solvation model which integrates a subtractive ONIOM(QM:SQM) [3] description of the solute into the EC-RISM formalism, combining established high-level QM methodology with EMPIRE [4] for the low-level semiempirical (SQM) component. We show how the set of equations used within this model can be derived, by similar approximations to the ones used for the ONIOM-PCM solvation model. [5] Extending previous schemes, [1] we develop an empirical correction to the solute's Gibbs energy, which is free of any ONIOM partitioning error. The resulting model is then validated for the SAMPL6 pK_a challenge dataset. [1,6] Our promising results show that the novel ONIOM-EC-RISM approach ranks equal in prediction quality with our previous full-QM EC-RISM methodology, while simultaneously drastically reducing the required computational cost. Thus, the model offers a step forward for modelling systems in solution up to biomolecular targets.

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